

Total No. of Questions :6]

SEAT No. :

P2865

[5023]-103

[Total No. of Pages : 5

M.Sc. - I

ORGANIC CHEMISTRY

**CHO - 150 : Organic Reaction Mechanism & Stereochemistry
(2013 Pattern) (5 Credits) (Semester - I)**

Time : 3 Hours]

[Max. Marks : 50

Instructions to the candidates:

- 1) All questions are compulsory.*
- 2) Figures to the right indicate full marks.*
- 3) Answers to the two sections should be written in separate answer books.*

SECTION - I

Q1) Attempt any three of the following: **[9]**

- a) Cyclooctatriene is non - aromatic. Explain.
- b) $\text{Et-S-CH}_2\text{-CH}_2\text{-Cl}$ undergoes hydrolysis 10^7 times faster than $\text{EtO-CH}_2\text{-CH}_2\text{-Cl}$. Explain.
- c) Pyrrole exhibits acidic character, whereas pyridine shows basic character. Explain.
- d) Cis 2-amine cyclohexanol as reaction with nitrous acid give only a single product, while trans isomer gives two products. Explain.

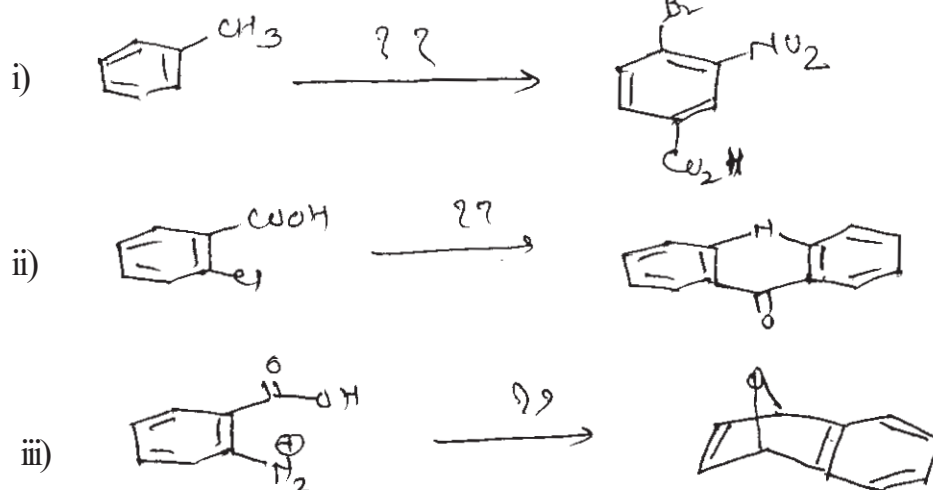
Q2) a) Write short note on (Any two) **[4]**

- i) Benzyne intermediate.
- ii) Non - Classical Carbocation.
- iii) Ipso Substitution reaction.

P.T.O.

b) Write reagents for following conversions (Any Two)

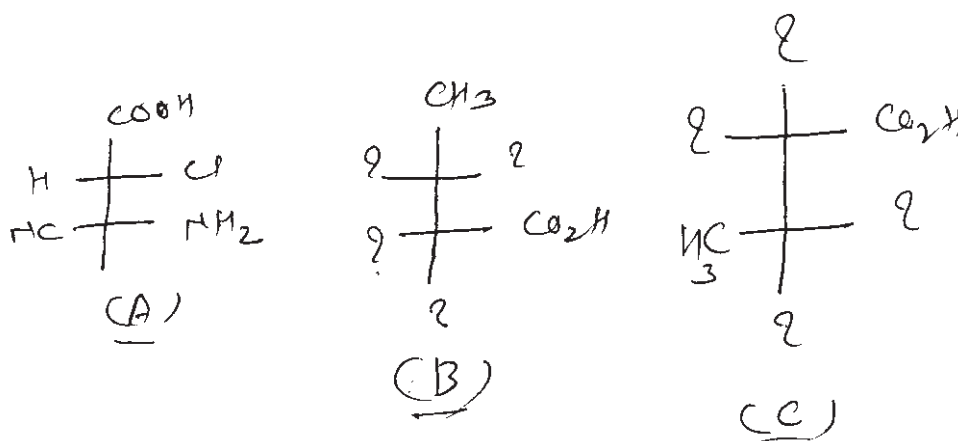
[4]



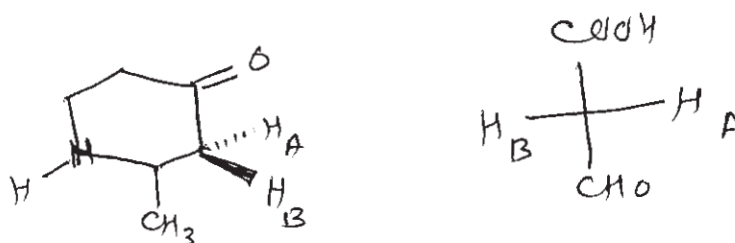
Q3) Attempt any four of the following:

[8]

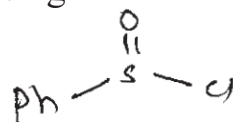
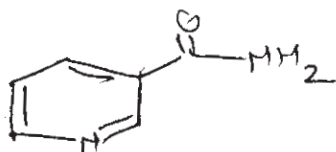
a) Write the equivalent structures B & C of the compound 'A'.



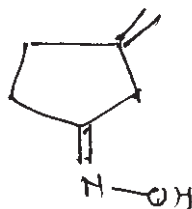
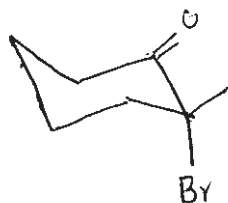
b) Assign Pro-R and Pro-S labels to H_A & H_B



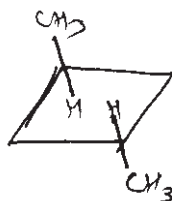
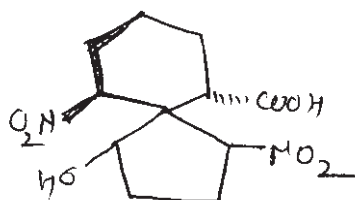
c) Assign Re & Si faces to the following



d) Assign R/S and E/Z configuration to the following.



e) Write the optical activity of the following.

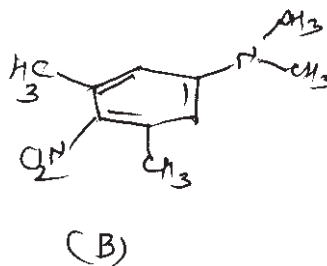
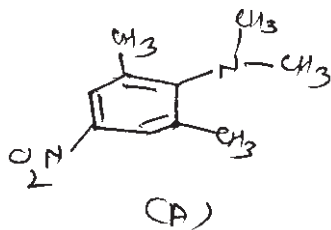


SECTION - II

Q4) Attempt any three of the following:

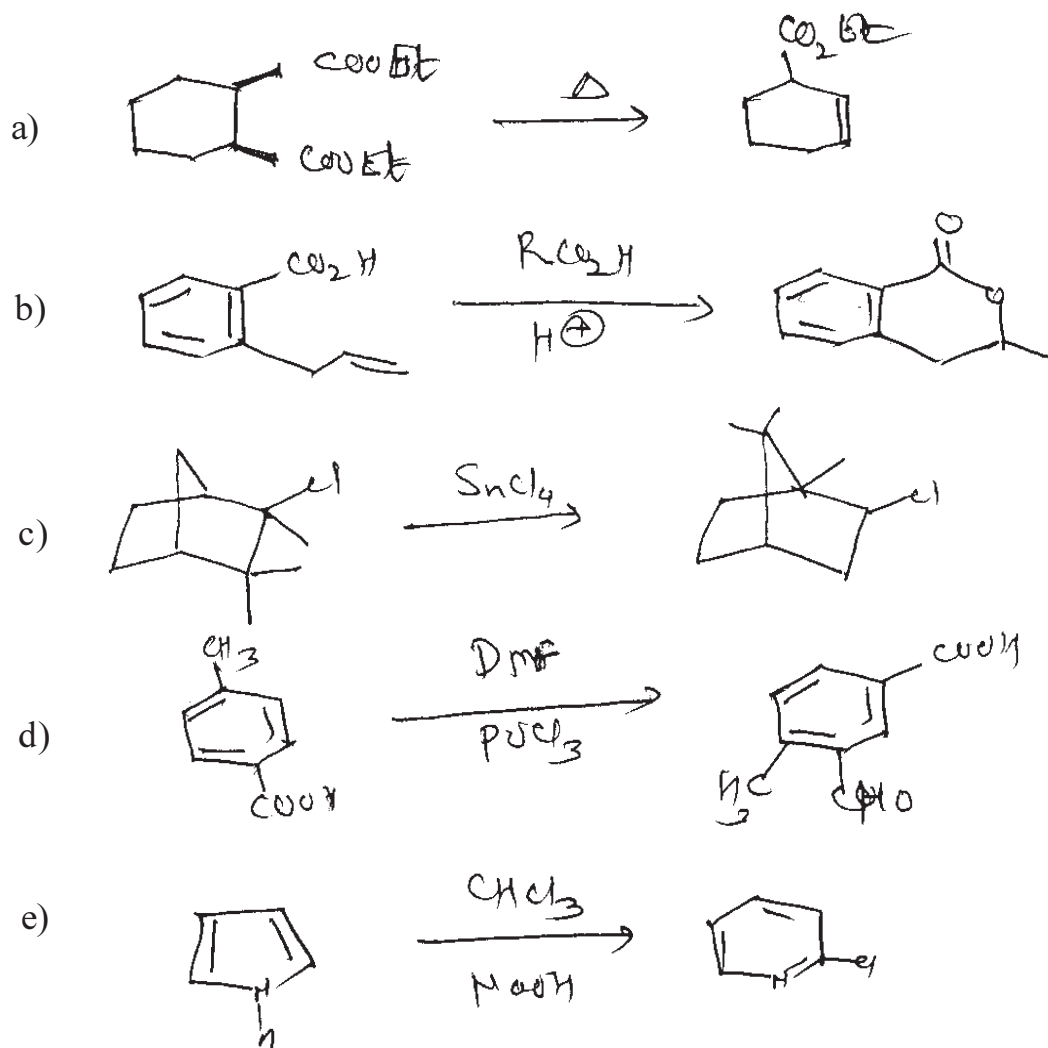
[9]

- Vinyl chloride resists hydrolysis, while allyl chloride is readily hydrolysed by S_N1 mechanism.
- Explain E, cb mechanism with suitable example.
- Nucleophilicity of Iodine, bromine & chlorine is in the order of $I^- > Br^- > Cl^-$ in protic solvent, where as the situation is reversed in a protic solvent. Why?
- Which of the following base is strong? Why?



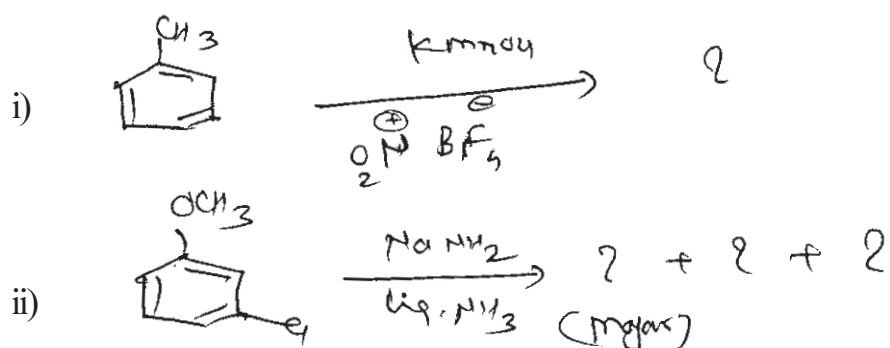
Q5) Suggest mechanism for the following (Any Four)

[8]



Q6) a) Predict the product/s

[4]



b) Attempt any two of the following:

[4]

- i) Write short note on Hoffmann elimination reaction.
- ii) Explain S_N^i reaction with example.
- iii) The compound (A) does not undergoes elimination reaction. Explain.

