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SEAT No. :

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[5315] - 427

T.Y. B.Sc. CHEMISTRY (Semester - IV)
CH - 346 (D) : Environmental And Green Chemistry
(2013 Pattern) (Elective - II) (Paper - VI)

Time : 2 Hours]

[Max. Marks : 40

Instructions to the candidates:

- 1) All questions are compulsory.*
- 2) Figures to the right indicate full marks.*
- 3) Neat diagrams must be drawn wherever necessary.*

Q1) Answer the following:

[10]

- i) What is mean by 'Sludge Gas'.
- ii) What is mean by 'Soil Horizon'.
- iii) What is role of carrier gas in gaschromatography (GC)
- iv) Which is most commonly used detector in HPLC.
- v) Why carbon dioxide is I.R. active.
- vi) What is mean by 'Heat of Vaporisation'.
- vii) How much energy stored by C = O bond.
- viii) Define 'Screening'
- ix) Define 'Incineration'.
- x) Explain 'Green house effect'.

Q2) a) Explain any two of the following :

[6]

- i) Explain 'Green house gases'.
- ii) Explain in detail "Electrodialysis".
- iii) Explain 'Mackereth oxygen cell'.

P.T.O.

- (b) Write short notes on (any two) [4]
- i) Wind energy
 - ii) Fizzy Water
 - iii) Sanitary Land fills

- Q3)** Answer any two of the following : [10]
- i) Explain in detail Secondary or biological waste water treatment.
 - ii) Explain in detail principle and working of gas chromatography (GC)
 - iii) Give an account on various chemical processes in water.

- Q4)** a) Explain Acid - base and Ionic reactions in soil and PH of soil. [6]

OR

What is Nuclear energy. Explain typical Nuclear fission power plant.

- b) Write note on any one of the following : [4]
- i) Global Warming and climate changes.
 - ii) Metals in water

