

Total No. of Questions—12]

[Total No. of Printed Pages—4+2

**[3662]-143**

**S.E. (Electrical) (I Sem.) EXAMINATION, 2009**

**MATERIAL SCIENCE**

**(2008 COURSE)**

**Time : Three Hours**

**Maximum Marks : 100**

- N.B. :—** (i) Answers to the two Sections should be written in separate answer-book.
- (ii) Answer Q. No. 1 or Q. No. 2, Q. No. 3 or Q. No. 4, Q. No. 5 or Q. No. 6 from Section I.
- (iii) Answer Q. No. 7 or Q. No. 8, Q. No. 9 or Q. No. 10, Q. No. 11 or Q. No. 12 from Section II.
- (iv) Figures to the right indicate full marks.
- (v) Use of logarithmic tables, slide rules and electronic pocket calculator is allowed.
- (vi) Assume suitable data, if necessary.

**Physical Constants :**

- (1) Angstrom unit (AU) =  $1 \times 10^{-10}$  metres.
- (2) Boltzmann's constant ( $k$ ) =  $1.380 \times 10^{-23}$  joule-degree $^{-1}$ .
- (3) Dielectric constant of free space ( $\epsilon_0$ )  
=  $8.85 \times 10^{-12}$  farad-metre $^{-1}$ .
- (4) Charge on electron ( $e$ ) =  $1.601 \times 10^{-19}$  coulombs.
- (5) Mass of electron ( $m$ ) =  $9.107 \times 10^{-31}$  kg.
- (6) Permeability of free space ( $\mu_0$ ) =  $4\pi \times 10^{-7}$ .
- (7) Mass of proton ( $m_p$ ) =  $1.627 \times 10^{-27}$  kg.
- (8) Velocity of light ( $c$ ) =  $2.998 \times 10^8$  metre per second.
- (9) Electron volt (eV) =  $1.602 \times 10^{-19}$  joules.
- (10) Debye unit =  $3.33 \times 10^{-30}$  coulomb-metre.

**P.T.O.**

1. (a) Derive Clausius-Mossotti relation as applied to dielectric materials in static field. State clearly the assumptions made. [6]
- (b) The number of atoms in volume of one cubic metre of hydrogen gas is  $9.8 \times 10^{26}$ . The radius of the hydrogen atom is 0.53 AU. Calculate the polarizability and relative permittivity. [6]
- (c) What is meant by loss tangent as referred to polar dielectrics. Hence give its significance. [4]

Or

2. (a) With a neat sketch explain the construction and working of photo-voltaic cell. [6]
- (b) A parallel plate capacitor having an area of  $6.45 \times 10^{-4} \text{ m}^2$  and a plate separation of  $2 \times 10^{-3} \text{ m}$  across which a potential of 10 V is applied. If a material having dielectric constant of 6.0 is positioned with in the region between the plates, calculate : [6]
- (i) The capacitance
- (ii) The magnitude of charge stored on each plate
- (iii) The electric field density D
- (iv) The polarization P.
- (c) Write a note on Piezo-electricity. [4]
3. (a) State the properties and applications of : [6]
- (i) Transformer oil
- (ii)  $\text{SF}_6$  gas.

- (b) Define breakdown voltage in connection with dielectric materials. www.sppuonline.com  
Explain various factors affecting breakdown strength of liquid dielectric material. [6]
- (c) Discuss the insulating materials used for power and distribution transformer. [4]

Or

4. (a) State different mechanisms of breakdown in vacuum. Explain any one. [6]
- (b) What do you mean by fibrous insulating material. What is their major drawback ? How can it be overcome ? [6]
- (c) What is meant by Townsend's primary and secondary ionization coefficients ? [4]
5. (a) Define relative permeability. Show that the relative permeability  $\mu_r = 1 + \chi_m$ , where  $\chi_m$  is the magnetic susceptibility. [6]
- (b) Explain classification of magnetic material on the basis of distribution of dipole moments. [8]
- (c) Calculate hysteresis loss in a specimen of iron subjected to magnetization of 50 Hz. The weight of the specimen is 40 kg and its density is 7680 kg/m<sup>3</sup>. The hysteresis loop area is equivalent to 198 J/m<sup>3</sup>. [4]

6. (a) What is Curie temperature for ferromagnetic material ? Describe Curie-Weiss law. [6]
- (b) Differentiate between : [6]
- (i) Soft and hard magnetic materials
- (ii) Ferromagnetism and antiferromagnetism.
- (c) A magnetic field strength of  $\text{Fe}_2\text{O}_3$  is  $10^6$  A/m. If the susceptibility of  $\text{Fe}_2\text{O}_3$  at room temperature is  $1.25 \times 10^{-3}$ , calculate induced magnetization, induced field density and permeability  $\mu$ . [6]

## SECTION II

7. (a) State the properties and applications of : [12]
- (i) Eureka
- (ii) Tungsten
- (iii) Kanthal.
- (b) Annealed copper has resistivity  $17.2 \times 10^{-9}$  ohm-m at  $20^\circ\text{C}$ . With 2 atomic percent of nickel, the resistivity of alloy of copper and nickel becomes  $4.06 \times 10^{-8}$  ohm-m. With the addition of 3 percent atomic silver, the resistivity of alloy of copper and silver becomes  $1.98 \times 10^{-8}$  ohm-m. What will be the resistivity of copper alloy for addition of 0.3 atomic percent of nickel and 0.2 atomic percent of silver at  $20^\circ\text{C}$ . [4]

Or

8. (a) Why is carbon preferred for brushes in electric machines ? [4]

- (b) What are the groups into which solders are grouped? Give their applications. [4]
- (c) A 230 volt filament lamp dissipates 60 watt at 2700°C. Resistivity of filament material at 20°C is  $4.3 \times 10^{-6}$  ohm-cm and its temperature coefficient at 20°C is 0.005/°C. Calculate the length of filament at 20°C if its diameter at 20°C is 0.028 mm. [8]

9. (a) What are carbon nanotubes? Discuss their electrical, mechanical and vibrational properties. List some applications of carbon nanotubes. [10]
- (b) Write a short note on single electron transistor. [6]

Or

10. (a) Discuss in brief the concepts of energy bands in insulators, semiconductors and conductors. [10]
- (b) Write a short note on BN nanotubes. [6]
11. (a) With a neat connection diagram, explain a method for determining dielectric strength of transformer oil as per relevant IS code of practice. [10]
- (b) What is  $\tan \delta$  of a dielectric? Explain. Describe the method of measurement of  $\tan \delta$  of a dielectric by Schering bridge as per IS code of practice. [8]

12. (a) With a neat connection diagram, explain the method for measurement of dielectric strength of air as per relevant IS code of practice. What inferences will you draw from this test ? [10]
- (b) What is partial discharge of a dielectric ? Explain a method to determine the partial discharge of a dielectric solid in the laboratory. [8]

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