

Total No. of Questions : 10]

SEAT No. :

P3566

[5560]-510

[Total No. of Pages : 3

T.E. (Civil Engineering)

ENVIRONMENTAL ENGINEERING - I

(2015 Pattern) (End Sem.) (301011) (Semester - II)

Time : 2½ Hours]

[Max. Marks : 70

Instructions to the candidates:

- 1) Answer any 7 questions.
- 2) Figures to the right indicate full marks.
- 3) Your answer will be valued as a whole.
- 4) Assume suitable data, if necessary.

- Q1)** a) Determine equivalent noise level for the following noise level measurements in premises. [5]

Time	4.30 am-6.30 am	6.30 am-8.30 am	8.30 am-10.30 am	10.30 am-12.30 pm	12.30 pm-2.30 pm	2.30 pm-4.30 pm
Sound in (dB)	44	53	67	72	68	56

- b) Give note on followings: [2+3]
- i) Stable Atmosphere.
 - ii) Plume Behavior

- Q2)** a) What is per capita demand? Give the water requirement for average Indian town on per capita basis. [6]

- b) Forecast the population for the year 1961 & year 1971 from following census data by Arithmetical Increase method. [4]

Census Year	1931	1941	1951
Population	35000	36500	37650

- Q3)** a) A settling tank is designed for an overflow rate of 5000 lit/m²/hr. What percentage of particles of diameter (i) 0.08mm and (ii) 0.04mm will be removed in this tank? Assume suitable data. [6]

- b) Enlist the types of Aerators. And explain in details any one from it. [4]

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- Q4)** a) Explain the following terms: [5]
i) Coagulation.
ii) Surface overflow rate.
iii) Flocculator.
b) With neat sketch explain the components of Rapid sand Filter and the step by step procedure of back washing. [5]
- Q5)** a) What is meant by Coagulation? Explain any one coagulant along with chemical reactions. [3]
b) Discuss the followings: [3]
i) Detention Period.
ii) Surface Loading.
c) Explain how plain sedimentation is differing than sedimentation with coagulation. [4]
- Q6)** a) Calculate the amount of bleaching powder required in kg per day for 10 MLD of water. The filtered water exerts a chlorine demand of 0.6 mg/lit to leave residual chlorine of 0.2 mg/lit. Chlorine available from bleaching powder is 40% [5]
b) Discuss in detail about Lime soda process and Ion exchange process. [5]
- Q7)** a) Explain in detail about Chlorine Ammonia treatment and state its merits. [5]
b) Explain about followings: [5]
i) Sources of Fluorides.
ii) Electrodialysis.
- Q8)** a) Discuss the followings: [5]
i) Break Point Chlorination.
ii) Methods of disinfection.
b) With suitable sketch explain about Solar distillation technique. [5]

Q9) a) Tabulate the comparison of Continuous and intermittent system of water supply. [5]

b) Explain any three methods of Rain water harvesting. [5]

Q10) a) Discuss the following. [2+2+2]

i) Pressure in distribution system.

ii) Radial system of water distribution.

iii) Water leakage detection techniques.

b) Discuss the points on which total capacity of reservoir is depends. [4]

